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## DMAP-Catalyzed [2 + 4] Cycloadditions of Allenoates with N‑Acyldiazenes: Direct Method to 1,3,4-Oxadiazine Derivatives

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Ph<sub>H<sub>2</sub>C</sub>

**S** Supporting Information

[ABSTRACT:](#page-2-0) An efficient DMAP-catalyzed [2 + 4] cycloaddition of allenoates and N-acyldiazenes is reported. The reaction involved embedding three heteroatoms into a sixmembered ring and generated 1,3,4-oxadiazine derivatives in moderate to good yields.

I eterocyclic skeletons are found in many naturally coccurring compounds used in medicinal chemistry.<sup>1</sup> 1,3,4-Oxadiazine is one of the most important frameworks for a variet[y](#page-2-0) of bioactive molecules (Figure  $1$ ).<sup>2</sup> Although many



Figure 1. Selected examples of biologically active molecules Figure 1. Selected examples of biologically active molecules Scheme 1. Organic Base-Catalyzed Cycloadditions containing a 1,3,4-oxadiazine skeleton. <br>
Scheme 1. Organic Base-Catalyzed Cycloadditions <br>
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platforms for the synthesis of various heterocycles have been developed, only a few reports have concerned the synthesis of 1,3,4-oxadiazines, particularly via intermolecular cyclizations.<sup>3</sup> For this reason, the assessment of facile protocols for the efficient generation of 1,3,4-oxadiazines still poses [a](#page-2-0) considerable challenge.

Cyclization is an effective synthetic strategy, and it has been widely used to construct cyclic compounds. Most of the cyclizations can proceed in metallo- and organocatalytic systems. During the past decades, organic base-catalyzed intermolecular cycloaddition, especially for allenoate-based cycloaddition, has been proven as an efficient and environmentally benign method to afford cyclic products from easily available starting materials.<sup>4</sup> Accordingly, the electrophiles,<sup>5</sup> including alkenes,  $6 \text{ }$  imines, $7 \text{ }$  aldehydes, $8 \text{ }$  azomethine imines, $9 \text{ }$ ylid[e](#page-2-0)s, $10^{\circ}$  [a](#page-3-0)nd aziridines, $11^{\circ}$  reacted with allenoates to form a wide range of car[bo](#page-3-0)- and h[et](#page-3-0)erocycles. [T](#page-3-0)o continue to explor[e](#page-3-0) other [or](#page-3-0)ganic base-pro[mo](#page-3-0)ted cycloadditions, the expansion of the scope of electrophile for constructing new heterocycles is

DMAP (20 mol %) toluene, rt, 48 h

highly desirable. Recently, a desulfonylative  $[3 + 2]$  cycloaddition of allylic carbonates with arylazosulfones to pyrazoles in the presence of tertiary phosphine was developed.<sup>12</sup> In continuation of work on the pursuit of other annulations, N-acyldiazenes, an important class of diazene with disti[nct](#page-3-0)ive reactivity, which were often used in carbene-catalyzed cycloadditions with ketenes<sup>13</sup> and aldehydes,<sup>14</sup> have received more attention. To the best of our knowledge, few papers have reported the use of N-a[cy](#page-3-0)ldiazenes in o[rga](#page-3-0)nic base-catalyzed cycloadditions. Unlike reported  $[2 + 4]$  cycloadditions using  $\alpha$ , $\beta$ -unsaturated imines, ketones or aldehydes as electrophiles with only one heteroatom in the six-membered ring (Scheme 1, eqs 1 and  $2$ ,<sup>15</sup> the synthesis of three heteroatoms in a six-membered ring in one step is rare, especially for organic base-catalyzed an[nu](#page-3-0)lations. Herein, a DMAP-catalyzed  $[2 + 4]$  cycloaddition





Received: May 15, 2015 Published: June 10, 2015 is reported using N-acyldiazenes as electrophiles to mix three heteroatoms in a six-membered ring and generate 1,3,4oxadiazines (Scheme 1, eq 3).

The first investigation was conducted with benzyl allenoate (1a) and phenyl(p[he](#page-0-0)nyldiazenyl)methanone (2a) in the presence of 10 mol % of 4-(dimethylamino)pyridine (DMAP) as a catalyst (Table 1, entry 1). 1,3,4-Oxadiazine





 $a$ Reaction conditions: 1a (0.30 mmol), 2a (0.20 mmol), catalyst (0.04 mmol, 20 mol %), solvent  $(2.0 \text{ mL})$ , rt, 48 h.  $b^b$ Isolated yield. <sup>c</sup>10 mol % of DMAP.  $\frac{d_{30}}{d_{30}}$  mol % of DMAP.  $\frac{e_{\text{At 0}}}{d_{30}}$  °C.  $\frac{e_{\text{At 0}}}{d_{30}}$  °C.  $\frac{e_{\text{At 0}}}{d_{30}}$  °C. reaction occurred.  ${}^{h}ND$  = no desired product was detected.

derivative 3aa was produced in 61% yield via a  $[2 + 4]$ cycloaddition pathway. The product yields increased to 70% and 71% after the amount of DMAP was increased to 20 and 30 mol %, respectively (Table 1, entries 2 and 3). Further examination of the reaction temperature indicated that room temperature is the best choice for this  $[2 + 4]$  cycloaddition (Table 1, entries 4 and 5). The solvent effect was also examined by screening toluene, acetone, EtOAc,  $CH<sub>3</sub>CN$ ,  $CH_2Cl_2$ , THF, DMF, and DMSO (Table 1, entries 6–12), of which toluene was found to be the most suitable medium. Other tertiary amines, such as pyridine, DABCO,  $Et<sub>3</sub>N$ , and DBU, were also used as catalysts, giving either poor product yield or no reactions (Table 1, entries 13−16). Meanwhile, tertiary phosphine shut down the cycloaddition completely (Table 1, entries 17 and 18). Optimized reaction conditions were determined using 20 mol % of DMAP as a catalyst in toluene solution at room temperature for 48 h.

With the optimized conditions in hand, the scope of the reaction was expanded to other N-acyldiazenes (2), as illustrated in Scheme 2. The reactions preceded smoothly to give the desired 1,3,4-oxadiazines with good yields in most of the cases. For the monosubstituted  $Ar<sup>1</sup>$  groups, both electronrich and electron-poor groups were compatible with these reaction conditions, and no obvious substitution effect was



 $a$ Reaction conditions: 1a (0.30 mmol), 2 (0.20 mmol), DMAP (0.04 mmol), toluene  $(2.0 \text{ mL})$ , rt, 48 h.  $^{b}$ Isolated yield.

observed (3aa-ad,ag). For a NO<sub>2</sub>-substituted diazene involved in the reaction, no desired product 3ae was detected. Using  $2f$  with a Cl on the *ortho-position* of the  $Ar<sup>1</sup>$  group only afforded a trace amount of 3af, and this was ascribed to the steric effect (3af vs 3ac and 3ag). Disubstituted  $Ar^1$  groups were also tested, specifically 2,4-dichloro, 3,4-dichloro, and 3,5-dichloro, giving the corresponding products (3ah−aj) in 60−80% yields. Furthermore, changing substituents on the  $Ar^2$  group in substrate 2 was also found to be suitable for the reaction, affording the desired products (3ak−ap) in moderate to good yields, except 3aq and 3ar due to their strong electron-withdrawing and steric effects, respectively. Substrates with two or three electron-rich groups, such as 3,4 dimethoxy, 3,5-dimethoxy, and 3,4,5-trimethoxy, reacted with 1a to afford the corresponding products (3as−au) in 70−80% yields. For diazenes 2 having a heteroaromatic or naphthalen-1-yl group (Ar<sup>2</sup> ), the anticipated products (3av−ax) were isolated in 63−69% yields. The structure of 3al was unambiguously confirmed by a single-crystal X-ray analysis.<sup>16</sup>

To further evaluate the scope of this reaction, other γsubstituent allenoates and active diazenes were examined, a[nd](#page-3-0) the results are listed in Scheme 3. Allenoates with different γsubstituents, including methyl and ethyl penta-2,3-dienoates, reacted smoothly with variou[s](#page-2-0) diazenes to produce the

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 $a$ Reaction conditions: 1 (0.30 mmol), 2 (0.20 mmol), DMAP (0.04 mmol), toluene  $(2.0 \text{ mL})$ , rt, 48 h.  $^{b}$ Isolated yield.

corresponding 1,3,4-oxadiazines in modest yields. N-Acetyldiazene is an effective electrophile. It reacted with 1a, albeit producing 3ay in only 33% yield. However, reaction with Ncarbopropoxy-substituted diazene failed (3az).

A plausible mechanism was proposed for this  $[2 + 4]$ cyclization, as depicted in Scheme 4. The first step involves the activation of allene ester 1a by DMAP to generate a zwitterionic intermediate A. Subsequent γ-nucleophilic attack of the electrophile 2a provided B, which underwent intramolecular Michael addition of oxygen anion to carbon atom to produce intermediate C. Finally, the catalyst DMAP was eliminated, and the  $C=C$  double bond was regenerated, to afford the product 3aa.

In summary, a direct synthetic method for the preparation of 1,3,4-oxadiazins via a DMAP-catalyzed [2 + 4] cycloaddition of allenoates with N-acyldiazenes is reported here. The reactions generated the corresponding products with good yields in most cases under simple and mild reaction conditions. This organic base-catalyzed one-step cycloaddition reaction generated three heteroatoms in a six-membered ring, providing new synthetic protocols for further heterocyclic Scheme 4. Proposed Mechanism

 $PhH<sub>2</sub>C$ 





synthesis. Further development of diazenes is currently underway.

### ■ ASSOCIATED CONTENT

#### **S** Supporting Information

Full experimental details and characterization data for all products. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/ acs.orglett.5b01237.

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#### Notes

The authors declare no competing financial interest.

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#### ■ REFERENCES

(1) Vitaku, E.; Smith, D. T.; Njardarson, J. T. J. Med. Chem. 2014, 57, 10257.

(2) (a) Dekeyser, M. A.; Mitchell, D. S.; Downer, R. G. H. J. Agric. Food Chem. 1994, 42, 1783. (b) Trepanier, D. L.; Krieger, P. E.; Eble, J. N. J. Med. Chem. 1965, 8, 802. (c) Shindy, H. A.; El-Maghraby, M. A.; Eissa, F. M. Dyes Pigments 2006, 70, 110. (d) Bakavoli, M.; Rahimizadeh, M.; Shiri, A.; Eshghi, H.; Vaziri-Mehr, S.; Pordeli, P.; Nikpour, M. Heterocycl. Commun. 2011, 17, 49. (3) (a) Matsubara, R.; Kobayashi, S. Angew. Chem., Int. Ed. 2006, 45, 7993. (b) Kristinsson, H.; Winkler, T.; Mollenkopf, M. Helv. Chim. Acta 1986, 69, 333. (c) Aly, A. A.; Ehrhardt, S.; Hopt, H.; Dix, I.; Jones, P. G. Eur. J. Org. Chem. 2006, 335. (d) Milcent, R.; Barbier, G. J. Heterocycl. Chem. 1992, 29, 1081. (e) Ismail, S. M. M.; Baines, R. A.; Downer, R. G. H.; Dekeyser, M. A. Pestic. Sci. 1996, 46, 163. (f) Bassam, F.; Jones, R. G. Chem. Commun. 1979, 917. (g) Tiecco, M.; Testaferri, L.; Marini, F. Tetrahedron 1996, 52, 11841. (h) Jones, R. A.; Gonzalez, B. A.; Arques, J. S.; Pardo, J. Q.; King, T. J. J. Chem. Soc., Perkin Trans. 1 1984, 1423.

(4) For selected reviews about organic base-catalyzed intermolecular cycloadditions, see: (a) Fan, Y. C.; Kwon, O. Chem. Commun. 2013, 49, 11588. (b) Zhao, Q.-Y.; Lian, Z.; Wei, Y.; Shi, M. Chem. Commun. 2012, 48, 1724. (c) Ye, L.-W.; Zhou, J.; Tang, Y. Chem. Soc. Rev. 2008, 37, 1140. (d) Denmark, S. E.; Beutner, G. L. Angew.

Chem., Int. Ed. 2008, 47, 1560. (e) Nair, V.; Menon, R. S.; Sreekanth, A. R.; Abhilash, N.; Biju, A. T. Acc. Chem. Res. 2006, 39, 520. (f) Lu, X.; Du, Y.; Lu, C. Pure Appl. Chem. 2005, 77, 1985. (g) Methot, J. L.; Roush, W. R. Adv. Synth. Catal. 2004, 346, 1035. (h) Valentinejr, D. H.; Hillhouse, J. H. Synthesis 2003, 317. (i) Lu, X.; Zhang, C.; Xu, Z. Acc. Chem. Res. 2001, 34, 535.

(5) Wang, Z.; Xu, X.; Kwon, O. Chem. Soc. Rev. 2014, 43, 2927. (6) For selected cycloaddition examples using alkenes as electrophiles, see: (a) Li, E.; Huang, Y. Chem.—Eur. J. 2014, 20, 3520. (b) Chen, X.-Y.; Wen, M.-W.; Ye, S.; Wang, Z.-X. Org. Lett. 2011, 13, 1138. (c) Steurer, M.; Jensen, K. L.; Worgull, D.; Jørgensen, K. A. Chem.—Eur. J. 2012, 18, 76. (d) Du, Y.; Lu, X.; Zhang, C. Angew. Chem., Int. Ed. 2003, 42, 1035. (e) Lu, X.; Lu, Z.; Zhang, X. Tetrahedron 2006, 62, 457. (f) Guan, X.-Y.; Shi, M. J. Org. Chem. 2009, 74, 1977. (g) Zheng, J.; Huang, Y.; Li, Z. Org. Lett. 2013, 15, 5758. For other related examples, see: (h) Jia, S.; Su, S.; Li, C.; Jia, X.; Li, J. Org. Lett. 2014, 16, 5604. (i) Su, S.; Li, C.; Jia, X.; Li, J. Chem.Eur. J. 2014, 20, 5905. (j) Li, J.; Wang, N.; Li, C.; Jia, X. Chem.-Eur. J. 2012, 18, 9645.

(7) For selected cycloaddition examples using imines as electrophiles, see: (a) Li, E.; Jia, P.; Liang, L.; Huang, Y. ACS Catal. 2014, 4, 600. (b) Zhao, H.; Meng, X.; Huang, Y. Chem. Commun. 2013, 49, 10513. (c) Shi, Y.-L.; Shi, M. Org. Lett. 2005, 7, 3057. (d) Meng, X.; Huang, Y.; Zhao, H.; Xie, P.; Ma, J.; Chen, R. Org. Lett. 2009, 11, 991. (e) Wang, Y.-Q.; Zhang, Y.; Dong, H.; Zhang, J.; Zhao, J. Eur. J. Org. Chem. 2013, 3764. (f) Yang, L.-J.; Wang, S.; Nie, J.; Li, S.; Ma, J.-A. Org. Lett. 2013, 15, 5214.

(8) For selected cycloaddition examples using aldehydes as electrophiles, see: (a) Xu, S.; Zhou, L.; Ma, R.; Song, H.; He, Z. Chem.—Eur. J. 2009, 15, 8698. (b) Zhu, X.-F.; Henry, C. E.; Wang, J.; Dudding, T.; Kwon, O. Org. Lett. 2005, 7, 1387. (c) Zhu, X.-F.; Schaffner, A.-P.; Li, R. C.; Kwon, O. Org. Lett. 2005, 7, 2977. (d) Xu, S.; Zhou, L.; Ma, R.; Song, H.; He, Z. Org. Lett. 2010, 12, 544. (e) Wang, L.-F.; Cao, X.-P.; Shi, Z.-F.; An, P.; Chow, H.-F. Adv. Synth. Catal. 2014, 356, 3383.

(9) For selected cycloaddition examples using azomethine imines as electrophiles, see: (a) Na, R.; Jing, C.; Xu, Q.; Jiang, H.; Wu, X.; Shi, J.; Zhong, J.; Wang, M.; Benitez, D.; Tkatchouk, E.; Goddard, W. A., III; Guo, H.; Kwon, O. J. Am. Chem. Soc. 2011, 133, 13337. (b) Jing, C.; Na, R.; Wang, B.; Liu, H.; Zhang, L.; Liu, J.; Wang, M.; Zhong, J.; Kown, O.; Guo, H. Adv. Synth. Catal. 2012, 354, 1023.

(10) (a) Li, K.; Hu, J.; Liu, H.; Tong, X. Chem. Commun. 2012, 48, 2900. (b) Jia, Z.-J.; Daniliuc, C. G.; Antonchick, A. P.; Waldmann, H. Chem. Commun. 2015, 51, 1054.

(11) Guo, H.; Xu, Q.; Kwon, O. J. Am. Chem. Soc. 2009, 131, 6318. (12) Zhang, Q.; Meng, L.-G.; Wang, K.; Wang, L. Org. Lett. 2015, 17, 872.

(13) Huang, X.-L.; He, L.; Shao, P.-L.; Ye, S. Angew. Chem., Int. Ed. 2009, 48, 192.

(14) (a) Chan, A.; Scheidt, K. J. Am. Chem. Soc. 2008, 130, 2740. (b) Yang, L.; Wang, F.; Lee, R.; Lv, Y.; Huang, K.-W.; Zhong, G. Org. Lett. 2014, 16, 3872. (c) Taylor, J. E.; Daniels, D. S. B.; Smith, A. D. Org. Lett. 2013, 15, 6058.

(15) For organic base-catalyzed  $[2 + 4]$  cyclizations using  $\alpha$ , $\beta$ unsaturated aldehydes, imines, and ketones as substrates, see: (a) Shi, Z.; Loh, T.-P. Angew. Chem., Int. Ed. 2013, 52, 8584. (b) Chen, X.-Y.; Wen, M.-W.; Ye, S.; Wang, Z.-X. Org. Lett. 2011, 13, 1138. (c) Zhang, X.-C.; Cao, S.-H.; Wei, Y.; Shi, M. Org. Lett. 2011, 13, 5732. (d) Zheng, J.; Huang, Y.; Li, Z. Org. Lett. 2013, 15, 5064. (e) Wang, X.; Fang, T.; Tong, X. Angew. Chem., Int. Ed. 2011, 50, 5361. (f) Yao, W.; Dou, X.; Lu, Y. J. Am. Chem. Soc. 2015, 137, 54. (g) Ashtekar, K. D.; Staples, R. J.; Borhan, B. Org. Lett. 2011, 13, 5732. (h) Henry, C. E.; Kwon, O. Org. Lett. 2007, 9, 3069. (i) Dückert, H.; Khedkar, V.; Waldmann, H.; Kumar, K. Chem.—Eur. J. 2011, 17, 5130.

(16) The X-ray single crystal structure of 3al is available from the Cambridge Crystallographic Data Centre (CCDC 1058288).

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■ NOTE ADDED AFTER ASAP PUBLICATION

Scheme 1 was corrected on June 12, 2015.